



Amorphous TiO_2 Nanotubes as a Platform for Highly Selective Phosphopeptide Enrichment

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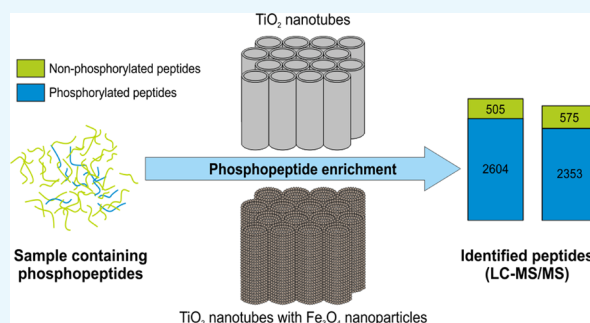
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Supporting Information

ABSTRACT: This work reports highly selective phosphopeptide enrichment using amorphous TiO_2 nanotubes (TiO_2NTs) and the same material decorated with superparamagnetic Fe_3O_4 nanoparticles ($\text{TiO}_2\text{NTs@Fe}_3\text{O}_4\text{NPs}$). TiO_2NTs and $\text{TiO}_2\text{NTs@Fe}_3\text{O}_4\text{NPs}$ materials were applied for phosphopeptide enrichment both from a simple peptide mixture (tryptic digest of bovine serum albumin and α -casein) and from a complex peptide mixture (tryptic digest of Jurkat T cell lysate). The obtained enrichment efficiency and selectivity for phosphopeptides of TiO_2NTs and $\text{TiO}_2\text{NTs@Fe}_3\text{O}_4\text{NPs}$ were increased to 28.7 and 25.3%, respectively, as compared to those of the well-established TiO_2 microspheres. The enrichment protocol was extended for a second elution step facilitating the identification of additional phosphopeptides. It further turned out that both types of amorphous TiO_2 nanotubes provide qualitatively new physicochemical features that are clearly advantageous for highly selective phosphopeptide enrichment. This has been confirmed experimentally resulting in substantial reduction of non-phosphorylated peptides in the enriched samples. In addition, $\text{TiO}_2\text{NTs@Fe}_3\text{O}_4\text{NPs}$ combine high selectivity and ease of handling due to the superparamagnetic character of the material. The presented materials and performances are further promising for applications toward a whole range of other types of biomolecules to be treated in a similar fashion.



INTRODUCTION

Nanomaterials undoubtedly represent the most investigated group of materials. Many nanomaterials have already found major applications in life science and biomedical devices. In particular, functionalized nano/microstructures are not only widely used in biomedical and biochemical research but also gradually becoming an integral part of many diagnostic or therapeutic tools used in clinical practice.^{1,2} Some of these materials have been fully exploited in the analysis of biological samples, especially for ultrahigh-throughput and parallel analysis at single-molecule levels. For example, in proteomics, nanomaterials are applied in the preanalytical enrichment phase combined with subsequent analysis of posttranslational modifications of proteins/peptides.³

Phosphorylation is a reversible covalent modification of proteins/peptides through which the protein/peptide function is regulated in response to extracellular stimuli.⁴ Because phosphate groups carry highly negative charges, the phosphorylation of a protein can substantially change its conformation,

which further affects the biological function, stability, and/or localization of the protein as well as its interaction with other proteins.⁵ One of the major goals in phosphoproteomics, the field that studies the dynamic changes of phosphoproteins over time, is to identify and describe the relationship of the phosphorylation regulation with the mechanism and pathogenesis of many serious diseases.^{6–8}

The analysis of phosphopeptides by mass spectrometry has many pitfalls that often distort the results and even lead to wrong conclusions. The low abundance of phosphopeptides, their low ionization efficiency, the labile phosphate moiety, and the interference of non-phosphorylated peptides strongly limit the comprehensive characterization of protein phosphorylation using mass spectrometry.^{9–11} These problems are often addressed by a three-step approach: (i) enrichment of

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phosphopeptides and (ii) fractionation of phosphopeptides by liquid chromatography, which is then (iii) coupled with tandem mass spectrometry (LC–MS/MS).¹² Over the past years, many approaches, including the commonly used immobilized metal ion affinity chromatography (IMAC)^{13,14} and metal oxide affinity chromatography (MOAC),^{15–17} have been developed in order to specifically enrich phosphopeptides before their analysis by LC–MS/MS. The principle of the phosphopeptide enrichment by IMAC is based on the high affinity of the negatively charged and hydrophilic phosphate group to metal ions (e.g., Fe^{3+} and Ga^{3+}) immobilized on a chelating iminodiacetic or nitrilotriacetic acid. Although this method is selective and generally sensitive enough, it still suffers from a lower affinity for some phosphopeptides and is too susceptible to sample contaminants, for example, nucleic acids.¹⁴ In addition, the non-specific adsorption of acidic peptides and peptides containing histidine may reduce the specificity of phosphopeptide enrichment.¹⁸ While conventional immobilized metal ions in IMAC are mainly trivalent ($\text{Fe}^{3+}/\text{Al}^{3+}/\text{Ga}^{3+}$), tetravalent IMAC utilizes $\text{Zr}^{4+}/\text{Ti}^{4+}/\text{Ce}^{4+}$ ions and commonly a phosphonate binding chemistry.^{19–21} Like the TiO_2 -based one, Ti^{4+} –phosphonate IMAC has a high tolerance toward acidic buffers to allow the enrichment of more basic phosphopeptides.²² Nevertheless, IMAC utilizing Ti^{4+} has lower affinity toward the multiply phosphorylated peptides, as compared to conventional Fe^{3+} -IMAC enrichment.²³

In the enrichment of phosphopeptides by metal oxides, the specificity has been substantially improved by introducing hydroxy acids such as 2,5-dihydroxybenzoic acid (DHB),¹⁵ glutamic acid,²⁴ or lactic acid²⁵ in the loading and washing buffer. Since the introduction of effective phosphopeptide enrichment using DHB as the excluder for TiO_2 microspheres,¹⁵ many metal oxides have been successfully applied in the enrichment of phosphopeptides, either as metal oxides (ZrO_2 , Fe_3O_4 , Al_2O_3 , SnO_2 , HfO_2 , or CeO_2)^{26–31} or composite materials (e.g., TiO_2 – ZrO_2 and CeO_2 – ZrO_2 composites).^{32,33} The review of Wang et al.³ provides a well-organized summary of materials developed and applied for phosphorylated protein/peptide enrichment. Despite this progress, however, the development of metal oxide-based affinity materials for the specific enrichment of phosphopeptides is still an ongoing process. Some of the reported studies only tested simple model mixtures of defined proteins rather than real complex samples like cell lysates.^{3,32–34} Therefore, it is necessary to acquire suitable materials and to design a protocol that could effectively treat also the real complex peptide mixtures often analyzed in various biological studies.

A very promising example of material that could be very effective for the purification represents self-organized TiO_2 nanotubes prepared by anodization of titanium. These nanotubes possess many intriguing properties, such as their unique 1D structure, tunability of dimensions, stability, biocompatibility, and easy preparation.^{35,36} Consequently, they have been applied in many different bioapplications for the photocatalytic killing of cancer cells,³⁷ drug delivery,³⁸ biosensing,³⁹ protein adsorption and separation,^{40,41} and His-tagged protein purification.⁴² Surprisingly, the potential of TiO_2 nanotubes for phosphopeptide enrichment was only noticed recently, starting with the introduction of tunable TiO_2 nanotube arrays either directly incorporated into the microchip for on-chip enrichment of serum phosphopeptides⁴³ or nanotubes grown on a titanium wire.⁴⁴ The materials can be

further made magnetically active by utilizing the superparamagnetic properties of magnetic components (e.g., Fe_3O_4 , $\gamma\text{-Fe}_2\text{O}_3$).^{42,45} Consequently, these composite materials can be rapidly isolated with an external magnetic field and easily dispersed, greatly facilitating the separation procedure by reducing the loss of affinity materials and biological samples, factors that directly impact the enrichment efficiency.³ Magnetic affinity composites are usually core–shell structures where the magnetic core does not interact with the surrounding environment. Anodic TiO_2 nanotubes decorated by Fe_3O_4 nanoparticles represent an interesting composite material that can be easily manipulated using an external magnetic field and possesses unique features usable for various bioapplications. We have recently shown that this material can be directly engaged in affinity interactions for the isolation of His-tagged recombinant proteins.⁴² However, this material has great potential for the treatment of other biomolecules, which is demonstrated in the presented study in the case of phosphopeptides.

In this work, we introduce amorphous TiO_2 nanotubes (TiO_2NTs) and amorphous TiO_2 nanotubes decorated with superparamagnetic Fe_3O_4 nanoparticles ($\text{TiO}_2\text{NTs}@ \text{Fe}_3\text{O}_4\text{NPs}$) as innovative materials for the enrichment of phosphopeptides. The tryptic digests of standard proteins as well as of highly complex cell lysates were utilized for the demonstration of phosphopeptide enrichment performance of the TiO_2NTs and $\text{TiO}_2\text{NTs}@ \text{Fe}_3\text{O}_4\text{NPs}$ materials. In addition, we developed a two-step sequential elution procedure resulting in an increased number of the detected phosphopeptides for all tested materials.

MATERIALS AND METHODS

Preparation of TiO_2NTs and $\text{TiO}_2\text{NTs}@ \text{Fe}_3\text{O}_4\text{NPs}$.

TiO_2NTs and $\text{TiO}_2\text{NTs}@ \text{Fe}_3\text{O}_4\text{NPs}$ materials were prepared as described in our previous work.⁴² Commercially available TiO_2 microspheres with a nominal diameter of 10 μm (Titansphere, GL Sciences) were used as a reference. Morphological characterizations of materials were performed using a field emission scanning electron microscopy (SEM) system (JSM 7500F, JEOL Ltd.). The structure of the samples was analyzed by diffraction analyses using an X-ray diffractometer (Empyrean, Panalytical) with $\text{Cu K}\alpha$ radiation. The specific surface area and the pore size distribution of TiO_2NTs , $\text{TiO}_2\text{NTs}@ \text{Fe}_3\text{O}_4\text{NPs}$, and TiO_2 microspheres were determined according to the N_2 adsorption isotherms. Measurements of adsorption isotherms of nitrogen (purity, 99.999 vol %) were carried out using an ASAP 2020 and evaluated using the MicroActive software (Micromeritics, USA). Specific surface areas were calculated from nitrogen adsorption isotherms using the BET approach.⁴⁶ Pore size distributions were determined by means of the BJH method⁴⁷ using the Harkins–Jura equation to calculate the adsorbed layer thickness.⁴⁸

Preparation of Model Phosphopeptide Mixtures.

Tryptically digested phosphoprotein α -casein and bovine serum albumin were used as a simple model peptide mixture. Both proteins were resuspended in 50 mM NH_4HCO_3 . A given amount of dithiothreitol (DTT) was added to reach the final concentration of 20 mM, and the solution was incubated for 30 min at 56 °C. Then, iodoacetamide (IAA) was added to the final concentration of 40 mM, and the solution was incubated for 30 min protected from light at RT. Alkylation was quenched by raising the DTT concentration to 40 mM.

The protein was digested with sequencing-grade modified trypsin (Promega, USA) at 37 °C overnight using an enzyme/substrate ratio of 1:50 (w/w).

For the preparation of the complex peptide mixture, Jurkat T cells were cultivated in RPMI-1640 medium with 10% (v/v) fetal bovine serum (FBS, Sigma-Aldrich, USA) at 37 °C/5% CO₂ and passaged every 2 or 3 days. The harvested cells were first treated with 1 mM pervanadate for 10 min at 37 °C (3.5×10^7 cells/mL),⁴⁹ then resuspended in an ice-cold hypotonic lysis buffer (10 mM Tris at pH 7.5 and 0.5 mM MgCl₂), and lysed using a glass Dounce homogenizer with a tight pestle (20 strokes). The protein concentration in the lysate was measured using a Micro BCA kit (Thermo Pierce, USA). The proteins were reduced by the addition of DTT (final concentration, 10 mM) and left for 1 h at 37 °C followed by alkylation with IAA (final concentration 20 mM) for 30 min at RT in the dark. An excess of IAA was quenched by the addition of DTT to the final concentration of 20 mM, and the reaction mixture was left for 15 min at RT. The proteins were digested by TCPK-treated trypsin from bovine pancreas (Sigma-Aldrich, USA) at an enzyme/substrate ratio of 1:50 (w/w) at 37 °C overnight. The peptides were desalted on Discovery DSC-18 SPE cartridges (500 mg/3 mL, Sigma-Aldrich, USA), eluted with 80% acetonitrile (ACN)/0.1% trifluoroacetic acid (TFA) and vacuum-dried.

Phosphopeptide Enrichment. Phosphopeptide enrichment was performed with either TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, or commercial TiO₂ microspheres. Phosphopeptide enrichment was carried out for both simple and complex peptide mixtures following the previously reported protocols^{25,50,51} and extended for the second elution step in order to obtain higher phosphoproteome coverage. To optimize the binding and elution conditions, tryptically digested α -casein (20 pmol) and bovine serum albumin (BSA, 200 pmol) were used as a simple model mixture, and the resulting protocol was performed also for the complex peptide mixture. The complex peptide mixture was enriched in duplicate, and each elution fraction was measured twice by nanoLC–MS/MS. Each material (3 mg of TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, or TiO₂ microspheres) was dispersed separately in 80% ACN/0.1% TFA (v/v) in Eppendorf tubes and washed with 80% ACN/5% TFA (v/v) containing 1 M lactic acid. Dried tryptically digested whole cell lysates of Jurkat T cells (500 μ g) were reconstituted in a loading buffer of 80% ACN/5% TFA (v/v) containing 1 M lactic acid, loaded onto the washed material for 30 min, and the supernatant was saved for a further re-enrichment step. Each material was then sequentially washed twice with 80% ACN/5% TFA (v/v) containing 1 M lactic acid, twice with 80% ACN/5% TFA (v/v), and once with 20% ACN/0.5% TFA (v/v). Phosphopeptides were eluted from the material using 1% NH₄OH (first elution step) followed by elution with 3% NH₄OH/45% ACN (v/v) containing 50 mM (NH₄)₂HPO₄ (second elution step). Supernatant fractions from TiO₂NTs and TiO₂NTs@Fe₃O₄NPs saved after the loading step were re-enriched using 3 mg of TiO₂ microspheres with the same protocol utilized for the raw lysate digest enrichment. The pH of all the elution fractions was reduced to 2–3 with 20% TFA. Prior to nanoLC–MS analysis, the elution fractions were desalted using custom-made reversed-phase microcolumns (length, 10 mm; ID, 250 μ m) packed with 2.6 μ m Kinetex EVO C18 resin (Phenomenex, USA) in FEP (fluorinated ethylene propylene) tubing (VICI AG, Switzerland) blocked with a small piece of Whatman glass microfiber

filter (GE Healthcare, USA). Samples eluted with a non-linear gradient with gradually increasing ACN content (2–40% ACN/0.1% TFA) from the desalting microcolumns were dried using a vacuum concentrator (Eppendorf, Germany) and frozen.^{52,53}

Mass Spectrometric Analysis. Mass spectrometric analysis of the simple mixture following enrichment was performed on the MALDI LTQ Orbitrap XL mass spectrometer (Thermo Fisher Scientific, USA) with DHB solution (10 g L⁻¹ in 50% ACN/0.1% TFA) with 1% H₃PO₄ as a matrix.¹⁵ Data were acquired in positive mode with full MS scan (800–4000 *m/z*) at 60,000 FWHM.

Peptides from the complex mixtures were first dissolved in 20 μ L of 2% ACN/0.1% TFA, and 1 μ L was analyzed using an UltiMate 3000 HPLC system (Dionex, USA) including a μ -Precolumn (300 μ m \times 5 mm, C₁₈PepMap 5 μ m 100 Å particles; Dionex, USA) connected to an analytical NanoEase column (100 μ m \times 150 mm, Atlantis C₁₈ 3 μ m 100 Å particles; Waters, USA). The separation was performed with a bilinear gradient of 5–45% ACN/0.1% TFA over 81 min under a flow rate of 360 nL min⁻¹ and UV detection set to 215 nm. The separation of peptides for nanoLC-MS/MS analysis was done using the UltiMate 3000 RSLC-nano HPLC system (Dionex, USA) with a trap column (75 μ m \times 20 mm) packed with 3 μ m Acclaim PepMap100 C18 particles and a separation column (75 μ m \times 150 mm) packed with 2 μ m Acclaim PepMap RSLC C18 particles. The separation was performed with a dual linear gradient using 3–44% ACN over 89 min under a flow rate of 300 nL min⁻¹ and analyzed with a Q Exactive system (Thermo Fisher Scientific, USA) in positive mode with full MS scan (350–1650 *m/z*) at 70,000 FWHM and with the top 12 precursors in MS/MS at 17,500 FWHM. Proteome Discoverer software (Thermo Fisher Scientific, USA, version 2.2.0.388) connected to the Mascot searching engine (version 2.4.1) was used for identification of MS/MS spectra. The parameters for Mascot database searching were as follows: instrument, ESI-QUAD-TOF; protein database, UniProt human reference proteome UP000005640 (July 5, 2018); enzyme, trypsin; maximum missed cleavage sites, 2; taxonomy, all entries; precursor mass tolerance, 15 ppm; fragment mass tolerance, 20 mmu; static modification, carbamidomethyl (C); dynamic modifications – oxidation (M), acetyl (protein N-term), Gln- > pyro-Glu (N-term Q), Glu- > pyro-Glu (N-term E), pyro-carbamidomethyl (N-term C), phospho (ST), phospho (Y).

The statistical language R (version 3.5.0)⁵⁴ with RStudio interface (version 1.1.447) and pre-programmed Excel (Microsoft, USA) worksheets were used for processing the results exported from Proteome Discoverer software. The MS proteomics data were deposited to the ProteomeXchange Consortium via the PRIDE partner repository with the dataset identifier PXD012749 (DOI: 10.6019/PXD012749).

RESULTS AND DISCUSSION

TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, and TiO₂ Microspheres for Phosphopeptide Enrichment. One of the developed materials, TiO₂NTs@Fe₃O₄NPs, was recently described and characterized by SEM, STEM, and magnetic measurements in our previous work.⁴² In this work, TiO₂ nanotubes without Fe₃O₄ nanoparticle loading (noted as TiO₂NTs) were also employed for the evaluation of phosphopeptide enrichment capabilities. For the characterization of the materials, X-ray diffraction (XRD), specific surface area measurements, and

SEM analyses were performed. Figure 1a,b shows SEM images of TiO₂ microspheres as the reference material for

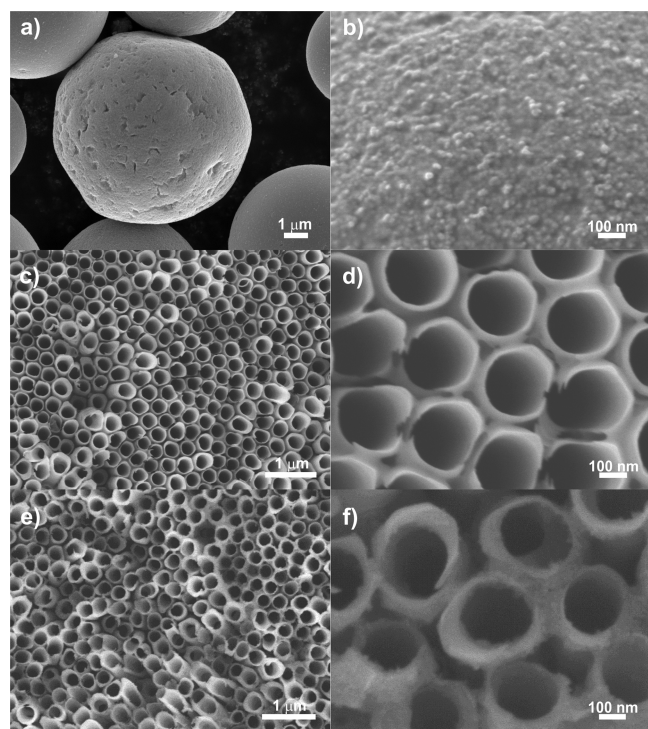


Figure 1. SEM images of (a, b) TiO₂ microspheres, (c, d) TiO₂NT layers, and (e, f) TiO₂NTs@Fe₃O₄NP layers at different magnifications.

phosphopeptide enrichment, and Figure 1c,d shows SEM images of the as-prepared amorphous TiO₂NTs with an internal diameter of ~ 230 nm at two different magnifications. In the next step, the TiO₂NTs were homogeneously decorated with magnetite nanoparticles to yield TiO₂NTs@Fe₃O₄NPs. Figure 1e,f confirms the presence of a homogeneous decoration of TiO₂NTs with magnetite nanoparticles, and this material was further used as a superparamagnetic material for phosphopeptide enrichment utilizing also the interactions of phosphopeptides with Fe₃O₄ nanoparticles on the surface of the composite TiO₂NTs@Fe₃O₄NP material.

To analyze the crystalline structure of tested materials, X-ray diffraction (XRD) analyses have been performed (Figure S-1). XRD patterns revealed that both types of TiO₂NTs remained amorphous after drying at 60 °C for 4 h and that commercial TiO₂ microspheres possess a tetragonal anatase crystalline structure.

The characterization of specific surface area (S_{BET}) using nitrogen adsorption/desorption isotherms of all three materials showed a substantial difference between both types of TiO₂NTs and the TiO₂ microspheres. The TiO₂ microspheres have a mesoporous structure, and S_{BET} was determined to be $127.5 \pm 0.8 \text{ m}^2 \text{ g}^{-1}$. This is a >5 -fold higher value as compared to that of TiO₂NTs@Fe₃O₄NPs for which S_{BET} was determined to be $24.2 \pm 0.3 \text{ m}^2 \text{ g}^{-1}$ and ~ 8.5 times higher than that for TiO₂NTs with $15.3 \pm 0.2 \text{ m}^2 \text{ g}^{-1}$. No porosity in the micro- and/or mesoporous range was observed for any TiO₂NT sample. According to the differences in the surface area of the materials, the maximum capacity for binding of phosphopeptides to TiO₂NTs and TiO₂NTs@Fe₃O₄NPs was

assumed to be lower, and thus the enrichment efficiency was further compared for all three materials.

Workflow and Principles of Phosphopeptide Enrichment. According to the literature, TiO₂ and Fe₃O₄ are capable of selective interaction with phosphopeptides;^{15,55} our TiO₂NTs consist of amorphous TiO₂, and they are strongly hydroxylated on the surface. Two additional interesting specific features are combined within TiO₂NTs@Fe₃O₄NPs: the unique morphology of strongly hydroxylated amorphous nanotubes along with the superparamagnetic property of the particles. Therefore, both materials should be capable of phosphopeptide adsorption and their enrichment from peptide mixtures. To verify that, two different samples were used to test the phosphopeptide enrichment by the TiO₂NTs and TiO₂NTs@Fe₃O₄NP materials: the simple peptide mixture containing digests of BSA and α -casein and the complex peptide mixture derived from whole cell lysates of Jurkat T cells. The overall enrichment performance of TiO₂NTs and TiO₂NTs@Fe₃O₄NP materials was also compared to the well-established phosphopeptide enrichment using TiO₂ microspheres.

Figure 2 schematically depicts the experimental workflow for the phosphopeptide enrichment using either TiO₂NTs or TiO₂NTs@Fe₃O₄NPs (Figure 2a) and TiO₂ microspheres as a reference (Figure 2b). The same separation conditions for the enrichment and effective isolation of singly and multiply phosphorylated peptides were essentially used for all three materials. To estimate the efficiency of phosphopeptide enrichment, the TiO₂ microspheres were also used for the re-enrichment of supernatant fractions resulting from enrichment of the complex peptide mixture with TiO₂NTs or TiO₂NTs@Fe₃O₄NPs. The second (additional) elution step was included in the protocol directly after the first elution step in order to obtain the phosphopeptides remaining on the material. The enrichment of the simple peptide mixture was assessed by MALDI-MS analysis, while for the complex peptide mixture, the nanoLC-MS/MS analysis was used. Separation of the TiO₂NTs and the TiO₂ microspheres from the liquid phase was done by centrifugation at 3000g, whereas TiO₂NTs@Fe₃O₄NPs were quickly and easily separated using a Nd magnet. The magnetic features of the TiO₂NTs@Fe₃O₄NPs allow faster and easier handling that is especially useful in parallel processing of several samples. Furthermore, the liquid–solid separation is more quantitative for the TiO₂NTs@Fe₃O₄NPs due to the low liquid retention in the test tube within replacement of individual solutions, which reduces the liquid carryover to the following steps.

Enrichment of Phosphopeptides from the Simple Peptide Mixture. To preliminarily evaluate the phosphopeptide enrichment ability of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs, the standard simple peptide mixture containing tryptic digests of α -casein and BSA in a molar ratio of 1:10 was utilized. Phosphopeptides originating from α -casein were enriched using the protocol described schematically in Figure 2a. The obtained (phospho)peptide mixture was directly analyzed using a MALDI-Orbitrap MS. The MALDI spectrum of the original peptide mixture (Figure 3a) contains almost no apparent signal of phosphopeptides, while the first eluents from the sample enriched with either TiO₂NTs (Figure 3b), TiO₂NTs@Fe₃O₄NPs (Figure 3d), or TiO₂ microspheres (Figure 3f) contain intensive signals of phosphopeptides. This confirms the ability of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs to enrich phosphopeptides from peptide mixtures in a similar way

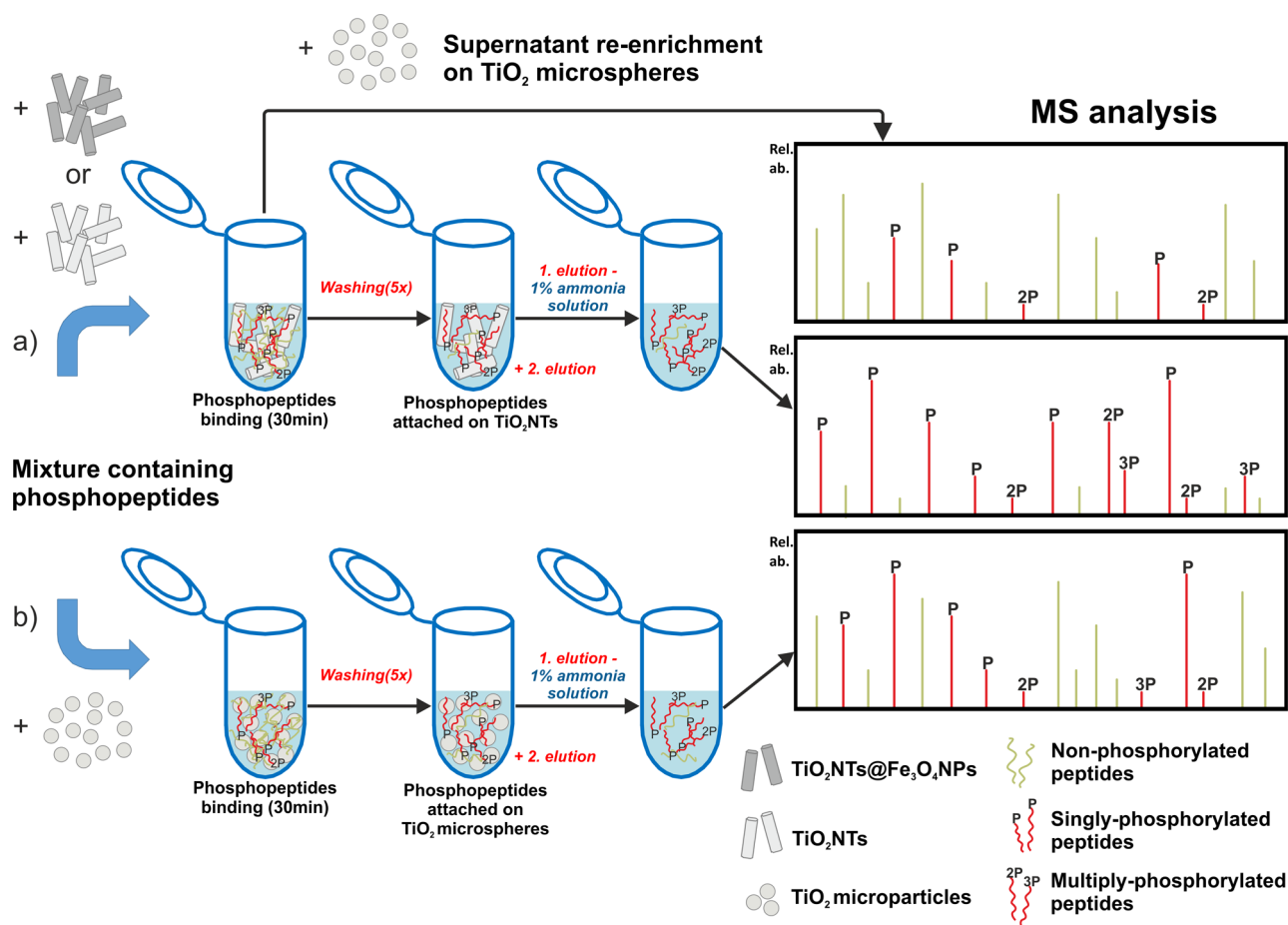


Figure 2. Schematic representation of the phosphopeptide enrichment workflow applied in this study using (a) TiO_2NTs or $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$ and (b) TiO_2 microspheres. Aliquots of tryptic digests of both standard proteins and complex cell lysates were used for the direct comparison of tested material.

as is usually done with TiO_2 microspheres. The amount of non-phosphorylated peptides observed in the first elution fraction is comparable for both TiO_2NTs or $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$ (Figure 3b,d) and TiO_2 microspheres (Figure 3f). The second elution fraction from TiO_2 microspheres (Figure 3g) contained a slightly higher abundance of non-specifically bound peptides (mainly in the range of m/z 700–1100) as compared to the second elution fraction from TiO_2NTs (Figure 3c) or $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$ (Figure 3e). Furthermore, the second elution fractions of all three materials (Figure 3c,e,g) showed a relative increase in signal abundance of multiply phosphorylated peptides. These differences between the first and the second elution fractions may be explained by the improved elution characteristics of the 3% $\text{NH}_4\text{OH}/45\%$ ACN (v/v) solution containing 50 mM $(\text{NH}_4)_2\text{HPO}_4$. This led us to further evaluate whether the second elution step could be possibly helpful in the analysis of multiply phosphorylated peptides. This might be important, for example, in studies focused on the analysis of hyperphosphorylated proteins related to certain pathology states.^{7,8}

Enrichment of Phosphopeptides from the Complex Peptide Mixture. In common real samples used for phosphopeptide profiling (cell lysates, extracellular matrix, tissue extracts, etc.), the phosphorylation modification often appears with a very low stoichiometry, which makes the phosphorylation analysis very difficult. Therefore, these samples need highly selective and specific methods for

phosphopeptide isolation in order to analyze the phosphopeptides successfully. Such methods should include both optimized sample processing conditions and suitable materials capable of handling the overwhelming sample complexity toward efficient phosphopeptide isolation. In this study, TiO_2NTs and $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$ were also tested for phosphopeptide isolation from the complex peptide mixture obtained as a tryptic digest of the whole cell lysate of Jurkat T cells treated with $\text{Na}_3\text{VO}_4/\text{H}_2\text{O}_2$.

For all materials, two enrichment replicates were analyzed, and nanoLC–MS/MS measurement of each replicate was performed in duplicate. Only peptides identified in both enrichment replicates and at least in one of two technical replicates with high confidence (1% FDR) were considered into the evaluation. Figure 4a shows a Venn diagram summarizing the number of phosphopeptides identified in the first elution fractions after phosphopeptide enrichment with TiO_2NTs , $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$, or TiO_2 microspheres. As can be seen, the majority of the identified phosphopeptides are identical for all three materials. This shows a similar performance of all three tested materials. Nevertheless, from the number of uniquely identified phosphopeptides for particular materials, it is obvious that the materials are partially complementary in the phosphopeptide coverage and that the use of more different materials for phosphopeptide enrichment can be beneficial to a deeper phosphopeptide analysis of the sample. Although it is known that the number of identified

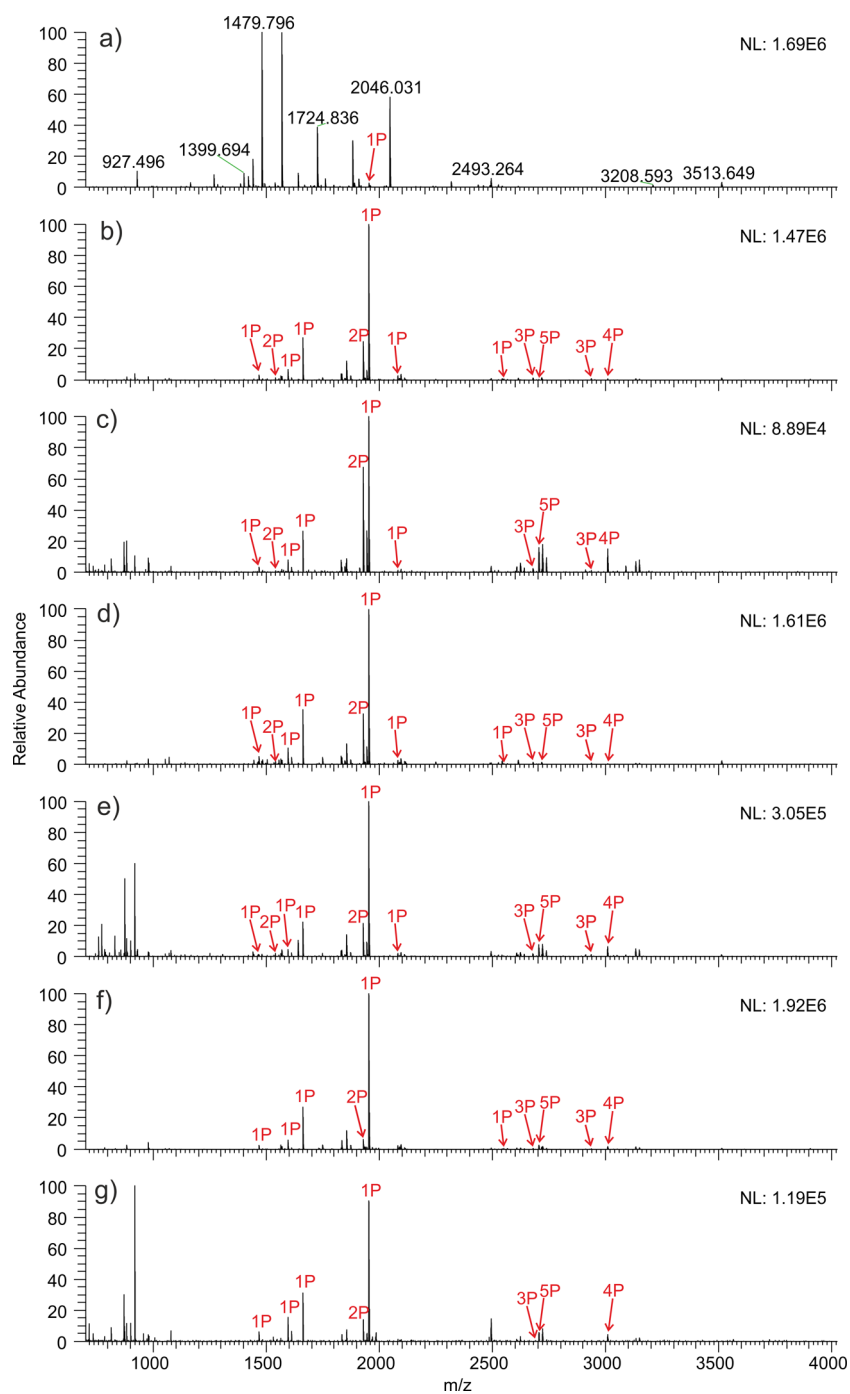


Figure 3. MALDI-Orbitrap mass spectra of tryptic peptides from α -casein and BSA. (a) Original peptide mixture, (b, c) first and second elution fraction after enrichment with TiO_2NTs , (d, e) first and second elution fraction after enrichment with $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$, and (f, g) first and second elution fraction after enrichment with TiO_2 microspheres. Phosphopeptides are marked with the number of phosphates (P). Oxidized forms of marked phosphopeptides and in-source loss of phosphate groups from this phosphopeptides are not highlighted in the spectra to keep the figure simple.

phosphopeptides can be increased by multiple sample injections in LC–MS analysis (22.1 and 22.7% more phosphopeptides for two sample injections of TiO_2NTs and $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NP}$ phosphopeptide isolates, respectively), the substitution of a second sample injection in the case of both nanotube-based materials by injection of a phosphopeptide sample obtained with TiO_2 microspheres led to a slightly increased phosphopeptide identification rate (27.9 and 36.1% for TiO_2NTs and $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$, respectively).

Figure 4b provides a summary overview of all identified phosphorylated and non-phosphorylated peptides in the first elution fraction for all materials. The total number of identified phosphopeptides was similar for both TiO_2NTs and TiO_2 microspheres (2604 and 2595, respectively), and it was slightly lower for $\text{TiO}_2\text{NTs}@Fe_3O_4\text{NPs}$ (2353, i.e., $\sim 9.3\%$ less phosphopeptides as compared to TiO_2 microspheres). However, a significant difference appeared between nanotube-based materials and TiO_2 microspheres in the number of

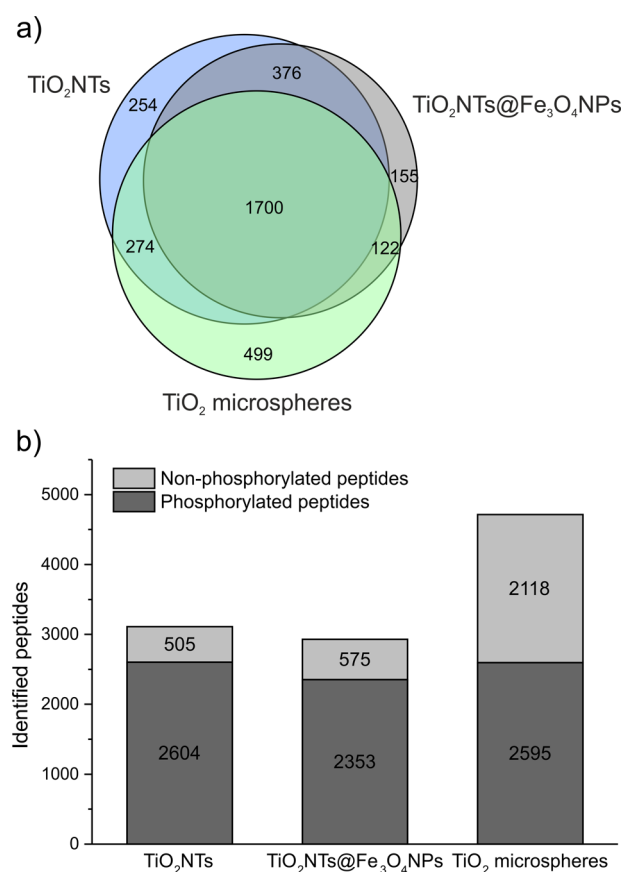


Figure 4. Comparison of the first elution fractions originating from the raw lysate digest enrichment using TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, and TiO₂ microspheres. (a) Venn diagram of identified phosphopeptides for particular materials and (b) column diagram with total numbers of identified phosphorylated and non-phosphorylated peptides.

identified non-phosphorylated peptides in the first elution fractions. Here, TiO₂NTs and TiO₂NTs@Fe₃O₄NP materials showed a significantly lower proportion of non-phosphorylated peptides in all identified peptides (505 out of 3019, 16.2%, and 575 out of 2928, 19.6%, respectively), as compared to TiO₂ microspheres (2118 out of 4713, 44.9%). This clearly shows a significantly improved selectivity of TiO₂NTs and TiO₂NTs@Fe₃O₄NP materials compared to the TiO₂ microspheres for the phosphopeptide enrichment. The decrease of non-phosphorylated peptides in elution fractions has a positive impact on the enriched sample purity obtained mainly from highly complex samples containing phosphopeptides in low abundance. Increased non-specific sorption of non-phosphorylated peptides on TiO₂ microspheres was also obvious from nanoLC analysis with UV detection where a significantly higher UV response for the first elution fractions from TiO₂ microspheres was observed (Figure S-2). The improved selectivity of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs is presumably caused by the corresponding material's composition, its structure, or non-crystallinity affecting its interaction with the peptides. Although the capacity of the nanotube-based materials for phosphopeptide enrichment proves to be comparable with TiO₂ microspheres under the used experimental conditions, further optimization of TiO₂NTs or TiO₂NTs@Fe₃O₄NPs is possible for an increased number of identified peptides while maintaining the superior material selectivity. This optimization

includes in particular tailoring the aspect ratio of the nanotubes (i.e., the ratio between the tube length and inner diameter) and the surface hydroxylation.

Two-Step Phosphopeptide Elution and Analysis of Multiply Phosphorylated Peptides. MOAC phosphopeptide enrichment based on the use of TiO₂ is popular for its simplicity and relatively high specificity. However, multiply phosphorylated peptides are usually observed to a lower extent in the TiO₂-enriched peptide mixtures as compared to, for example, IMAC phosphopeptide enrichment. One assumption is that certain (especially multiply) phosphorylated peptides may not be eluted under standard elution conditions from TiO₂-based materials.⁵⁰ To address this point, an easy second elution step in addition to the standard phosphopeptide elution with ammonia solution was introduced to the enrichment protocol in this work. The composition of the second elution solution was given by an increased concentration of ammonia (3 vs 1% NH₄OH), the introduction of ACN (45%, v/v) to diminish the non-specific sorption of peptides on the material, and the addition of phosphate salt (50 mM (NH₄)₂HPO₄) for an increased competitive elution of phosphopeptides. This simple additional step follows the first elution step, and thus it does not have any effect on the results obtained with the standard elution procedure. The processing of the second eluent is almost the same as for the first with the exception of the vacuum drying step, included prior to the reversed-phase-based sample clean-up for the following nanoLC-MS/MS analysis.

The first test with the simple mixture of two standard proteins suggested that the relative abundance of multiply phosphorylated peptides may increase in the second elution fraction (Figure 3c,e,g). To verify these observations, the second elution step was also included in the analysis of complex peptide mixtures derived from Jurkat T cells for the compared materials. The UV records from nanoLC analysis for TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, and TiO₂ microspheres (Figure S-3–S-5, respectively) confirm that the total peptide amount is significantly lower than that obtained in the first elution step. However, the following nanoLC-MS/MS analysis still identified a significant amount of phosphopeptides also in the second elution fractions (1785, 1188, and 1633 for TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, and TiO₂ microspheres, respectively; Figure 5a). The vast majority of these identified phosphopeptides were also found in the nanoLC-MS/MS analysis of the first elution fraction as shown in the Venn diagrams in Figure 5b. Nevertheless, when compared to the number of phosphopeptides identified in the first elution fraction, 216, 204, and 265 phosphopeptides were newly identified for TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, and TiO₂ microspheres comprising an increase of 8.3, 8.7, and 10.2% in the overall phosphopeptide coverage, respectively.

The obtained data for all materials and both elution fractions were also evaluated according to the number of phosphate groups on the peptide backbone present in identified phosphopeptides, and the summary is presented in the Table 1. From this point, all three materials are comparable with a slightly lower performance of TiO₂NTs, whereas TiO₂NTs@Fe₃O₄NPs show the highest ratio of multiply phosphorylated peptides toward singly phosphorylated ones. The hypothesis about the increased presence of multiply phosphorylated peptides in the second elution fraction was partially confirmed for TiO₂NTs@Fe₃O₄NPs where the relative ratio of multiply

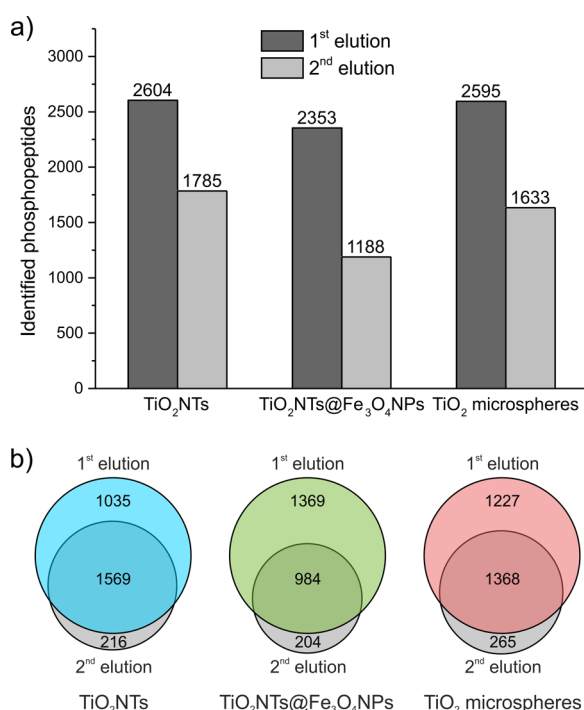


Figure 5. Comparison of the first and second elution fractions originating from the raw lysate digest enrichment using TiO₂NTs, TiO₂NTs@Fe₃O₄NPs, and TiO₂ microspheres. (a) Column diagram with total numbers of identified phosphopeptides and (b) Venn diagrams with intersections of both elution steps.

phosphorylated peptides was the highest among the materials and even increased in the second elution fraction.

Re-enrichment of Supernatant Fractions with TiO₂ Microspheres. To further characterize the phosphopeptide enrichment capabilities of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs, the supernatant fraction from the complex peptide mixture enrichment with nanotube-based materials was re-enriched using TiO₂ microspheres. The aim was to determine the level of effectiveness of phosphopeptide capture of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs and to test whether these materials are applicable to the enrichment of typical complex peptide mixtures originating from biological experiments. The results of these experiments are summarized in Figure S-6, which compares identified phosphopeptides and non-phosphorylated peptides both in the first elution fraction of raw lysate enrichment and in the first elution from the supernatant re-enrichment.

NanoLC reversed-phase analysis with UV detection showed a higher amount of peptides present in the elution fraction from the supernatant re-enrichment as compared to the first and second elution fractions from the raw lysate enrichment

(Figures S-3 and S-4). The following nanoLC-MS/MS analysis revealed that the supernatant fractions contained a lower number of identified phosphopeptides (905 and 1188 for TiO₂NTs and TiO₂NTs@Fe₃O₄NPs, respectively), whereas many of them (85.0 and 73.6%, respectively) have already been identified in the first elution fraction from the raw lysate enrichment. On the other hand, the re-enrichment of supernatant fractions originating from TiO₂NTs and TiO₂NTs@Fe₃O₄NPs using commercial TiO₂ microspheres led to the identification of other 136 and 194 phosphopeptides not identified previously in the first elution fraction from the raw lysate enrichment. These results show that the phosphopeptide enrichment using nanotube-based materials is very effective even though the results confirmed partial orthogonality between the TiO₂ microspheres, nanotube-based materials TiO₂NTs, and TiO₂NTs@Fe₃O₄NPs. This phenomenon has already been observed in the comparison with the identified phosphopeptides in the first elution fractions after raw lysate enrichment for all materials (Figure 4a). Then, this partial alternative selectivity can be also used in the re-enrichment of supernatant fractions in order to obtain higher phosphoproteome coverage of the analyzed sample although it is more laborious than the approach with the second elution step described in the previous section.

The results of the supernatant fraction re-enrichment show that the presumption of significantly lower capacity of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs was not true for these materials, sample amount, and conditions used in the enrichment protocol. The lower capacity was presumed based on specific surface area analysis showing the 8.5-fold and 5.3-fold lower surface area by material weight for TiO₂NTs and TiO₂NTs@Fe₃O₄NPs as compared to TiO₂ microspheres, respectively. Furthermore, in the re-enriched supernatant fractions obtained using TiO₂ microspheres, an ~8-fold higher amount of non-phosphorylated peptides was observed as compared with raw lysate enrichment (Figure S-6c). This is in line with the results presented in Figure 4 that show a lower selectivity of TiO₂ microspheres as related to nanotube-based materials, and it confirms the importance of selectivity in phosphopeptide enrichment for this type of sample. Obviously, the samples enriched with materials with a higher level of non-specific sorption (e.g., TiO₂ microspheres in this case) will result in an increased content of non-phosphorylated peptides in the elution fractions, further leading to the signal suppression of phosphopeptides in the MS analysis.

TiO₂NTs and TiO₂NTs@Fe₃O₄NPs differ from the commonly used crystalline TiO₂ microspheres in several aspects, which may be responsible for their different selectivities. In other studies with similar protocols for enrichment of phosphopeptides from complex mixtures, the overall selectivity of commercial crystalline TiO₂ microspheres

Table 1. Total Numbers of Identified Phosphopeptides in the Elution Fractions (EFs) Isolated from the Complex Peptide Mixture According to the Number of Phosphorylations per Peptide

sample	1 × phospho	2 × phospho	3 × phospho
TiO ₂ NTs – 1st EF	2464	140	0
TiO ₂ NTs – 2nd EF	1687	98	0
TiO ₂ NTs@Fe ₃ O ₄ NPs – 1st EF	2176	175	2
TiO ₂ NTs@Fe ₃ O ₄ NPs – 2nd EF	1087	101	0
TiO ₂ microspheres – 1st EF	2417	177	1
TiO ₂ microspheres – 2nd EF	1521	112	0

was found to be $\sim 70.1\%$ when DHB was used as an excluder⁵⁶ and 72.2% in the case of lactic acid, which served also for exclusion of non-phosphorylated peptides.⁵⁷ These numbers are higher than those observed for TiO₂ microspheres in this work (55.1%); however, both studies used diverse initial samples, a different sample-to-microsphere ratio, and a two-dimensional LC MS setup. Apart from crystalline microspheres, phosphopeptides can directly interact with the presented nanotubular materials not only via the available amorphous TiO₂ surface of nanotubes but also in the case of TiO₂NTs@Fe₃O₄NPs through the Fe₃O₄ nanoparticles attached on the nanotube TiO₂ walls. The binding characteristic of TiO₂ is described mainly as bidentate coordination of hydroxyl groups to positively polarized Ti(IV), which is responsible for the preferential capture and enrichment of phosphopeptides.^{58,59} In the IMAC technology, Ti⁴⁺ ions are utilized for phosphopeptide enrichment as well as Fe³⁺ ions.¹⁹ As part of the SIMAC protocol, the Fe³⁺ IMAC technology is also responsible for the improved enriching of the multiply phosphorylated peptides.⁵⁰ Moreover, bare Fe₃O₄ nanoparticles⁵⁵ and Fe₃O₄ nanoparticles incorporated in a polymer-based monolithic column⁶⁰ were previously utilized for the phosphopeptide enrichment. Compared to polymer compounds, inorganic supports, such as TiO₂, may be beneficial due to their stability in organic solvents (e.g., acetonitrile).⁴² TiO₂ as a support is even more useful since it can also enrich the phosphopeptides. Compared to other primarily crystalline materials available today, TiO₂NTs and TiO₂NTs@Fe₃O₄NPs possess an amorphous structure with a high number of hydroxyl groups on the surface responsible for very good hydrophilicity of the material and easy wetting of nanotube interiors.⁶¹

TiO₂-based nanotubes have already been described for phosphopeptide enrichment previously.^{43,44} However, the nanotubes presented by Wijeratne et al. differ in their crystalline structure because they were annealed and produced in the form of anatase, and therefore the efficacy of phosphopeptide enrichment was similar to commercial TiO₂ microspheres.⁴⁴ The inner nanotube diameter was also narrower, ~ 110 and ~ 45 nm.^{43,44} The inner diameter is important for accessibility of nanotube interiors to phosphopeptides; thus, the wider inner diameter (~ 230 nm) of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs and their amorphous character may be crucial for the phosphopeptide enrichment efficiency. On the other hand, the increased tube diameter means a lower surface/volume ratio and surface/weight ratio.

CONCLUSIONS

This paper reports the successful application of TiO₂NTs and TiO₂NTs@Fe₃O₄NP materials for phosphopeptide enrichment preferably suitable for complex biological samples. In addition, the improved enrichment protocol including a second elution step for more extensive identification of phosphopeptides is presented. By comparison of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs along with a reference material based on TiO₂ microspheres, it can be concluded that the amorphous TiO₂ nanotubular surface and the superparamagnetic version decorated with the Fe₃O₄ nanoparticles specifically interact with phosphopeptides. The performance of presented materials was comparable with well-established commercial TiO₂ microspheres although these possess a several times higher surface area by material weight. Our results unambiguously confirmed that the main difference

between nanotube-based materials and TiO₂ microspheres lies in a different proportion of the non-specific sorption. The amorphous character of TiO₂ nanotubes results in a significantly improved selectivity for phosphopeptide isolation and to an increased ratio of phosphopeptides to peptides in the enriched sample. Substantially improved selectivity of TiO₂NTs and TiO₂NTs@Fe₃O₄NPs for phosphopeptides is observed in the reduction of non-specific sorption of non-phosphorylated biomolecules, and the suppression of phosphopeptide signal within nanoLC-MS analysis is thus diminished. Furthermore, in the analysis of complex cell lysate digests, the introduction of the second elution step enhanced the overall phosphoproteome coverage in the sample. Finally, the added superparamagnetic properties of the TiO₂NTs@Fe₃O₄NPs are also important for the simplified sample handling during phosphopeptide enrichment.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsomega.9b00571.

Additional information about the materials' characteristics, nanoLC-UV analysis, and results of nano-LC-MS (PDF)

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Notes

The authors declare no competing financial interest.

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